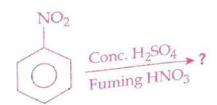


- 41. 50 ml of H₂ diffuses out through a small hole from a vessel in 20 mins. The time needed for 40 ml of O₂ to diffuse out from the same vessel is:
 - (A) 12 min
 - (B) 64 min
 - (C) 8 min 174242
 - (D) 32 min
- 42. Which of the following shows positive inductive effect?
 - (A) NO₂
 - (B) COOH /
 - (C) OCH₃
 - (D) -CN
- 43. What will be the product of the following reaction?



- (A) m-dinitrobenzene
- (B) o-dinitrobenzene
- (C) p-dinitrobenzene
- (D) Both o- and p-dinitrobenzene

44. The set of quantum numbers for 19th electron of Chromium atom is:

	n	1	m	8
(A)	3	0	0	1/2
(B)	3	2	-2	1/2
(C)	4	0	0	1/2
(D)	4	1	0	1/2

- **45.** Among the following covalent compounds, the compound having more polar character is :
 - (A) HI
 - (B) HCI
 - (C) HBr
 - (D) HF
- 46. The refining of Nickel metal is done by:
 - (A) Van Arkel Method
 - (B) Mond process
 - (C) Vapour-phase refining
 - (D) Zone refining .
- 47. When acetylene is passed through dil.H₂SO₄ in presence of HgSO₄ at 60°C, the organic compound formed is:
 - (A) Dimethylether ~
 - (B) Acetone
 - (C) Acetic acid
 - (D) Acetaldehyde

- 48. The ore containing two different metals
 - is:
 - (A) Haematite
 - (B) Galena
 - (C) Copper pyrite
 - (D) Magnetite
- **49.** Which of the following ion is the smallest in size ?
 - (A) , N^{3- ...}
 - (B) Na+
 - (C) F-
 - (D) O^{2-}
- of an ideal solution containing the non-volatile solute is equal to the mole fraction of the solute at a given temperature. This law is known as:
 - (A) Henry's law
 - (B) Van't Hoff's law
 - (C) Raoult's law
 - (D) Ostwald's dilution law /
- 51. The oxide of a metal contains 40% oxygen. If the valency of the metal is 3, its atomic mass will be:
 - (A) 8 9
 - (B) 16-
 - (C) 36
 - (D) 24

- 52. The volume of 0.05 N $\rm H_2SO_4$ solution needed to completely neutralise 25 ml of 0.1 N NaOH solution is :
 - (A) 25 ml /
 - (B) 50 ml
 - (C) 100 ml
 - (D) 12.5 ml
- 53. According to Le-Chatelier's principle maximum yield of ammonia is obtained at:
 - (A) High temperature and low pressure
 - (B) High pressure
 - (C) Low temperature
 - (D) Low temperature and High pressure
- 54. One drop of water weighs 0.018 g. Number of water molecules present in one drop of water is:
 - (A) 1×10^{-3}
 - (B) $6.02 \times 10^{20} \ \text{P}$
 - (C) 22.4×10^{-3}
 - (D) $6.02 \times 3 \times 10^2$

- 55. ZnS is not precipitated by passing H₂S through acidified ZnCl₂ solution, but CuS is precipitated by passing H₂S through acidified CuSO₄ solution. The reason for this is:
 - (A) K_{sp} CuS >> K_{sp} ZnS
 - (B) $K_{sp} CuS = K_{sp} ZnS$
 - (C) K_{sp} CuS << K_{sp} ZnS
 - (D) None of these /
- 56. The correct order of electron gain enthalpy among the following is:
 - (A) F > Cl > Br
 - (B) Br > C1 > F
 - (C) Cl > F > Br
 - (D) F > Br > Cl
- 57. According to VSEPR theory the shape of SF₆ molecule is:
 - (A) Trigonal bipyramidal
 - (B) Regular octahedral
 - (C) Pentagonal bipyramidal /
 - (D) Tetrahedral

- 58. The uncertainty in the momentum of an electron is 1×10^{-5} kg m/s. The uncertainty in its position will be $(h = 6.62 \times 10^{-34} \text{ kg m}^2/\text{s})$:
 - (A) 1.05×10^{-28} m
 - (B) $1.05 \times 10^{-26} \text{ m}$
 - (C) 5.27×10^{-30} m
 - (D) 5.27×10^{-28} m
- 59. Which of the following is an electrophile?
 - (A) H₂O /
 - (B) NH₃
 - (C) AlCl₃
 - (D) CH₃NH₂
- 60. In the standardisation of Na₂S₂O₃ using K₂Cr₂O₇ by iodometry, the equivalent mass of K₂Cr₂O₇ is:
 - (A) $\frac{\text{Mol.Mass}}{2}$
 - (B) $\frac{\text{Mol.Mass}}{3}$
 - (C) $\frac{\text{Mol.Mass}}{6}$
 - (D) Same as molecular mass

Where can the maximum biodiversity be

- 61. seen?
 - Tropical Rain Forest (A)
 - Temperate Rain Forest (B)
 - Alpine Forest (C)
 - Mediterranean Forest (D)
- Seedless vascular cryptogams are: 62.
 - Monocots (A)
 - Ferns (B)
 - Liverworts (C)
 - Gymnosperms (D)
- The maiden-hair tree is: 63.
 - (A) Adiantum
 - Pinus/ (B)
 - Ginkgo biloba (C)
 - Gnetum ula (D)
- The phylogenetic basis of classification 64. proposed by John Hutchinson was presented in the book:
 - Systema nature (A)
 - Introduction to the natural orders (B) of plants
 - Fundamental botanica (C)
 - The families of flowering plants (D)
- Which is not an element of phloem tissue? 65.
 - Sieve tubes (A)
 - Companion cells (B)
 - Bast fibres (C)
 - Wood parenchyma / (D)

- Sapwood is synonymous with: 66.
 - Bark (A)
 - Periderm (B)
 - Inner layers of secondary xylem-(C)
 - Outer layers of secondary xylem
- Indicate the features of a dicot stem: 67.
 - Vascular bundles closed, collateral and exarch
 - Vascular bundles scattered, more (B) towards the periphery
 - Vascular bundles with bundles (C) radial, exarch and closed
 - (D)/ Vascular bundles are conjoint, collateral, open and endarch
- Which are the products of cyclic 68. during photophosphorylation photosynthesis?
 - $ATP + O_2$ (A)
 - $NADH + O_2$ (B)
 - ATP + NADH (C)
 - ATP + NADH + O₂ (D)
- Which organelle does not participate in 69. photorespiration?
 - (A) Peroxisome
 - Mitochondria (B)
 - Golgi bodies (C)
 - Chloroplast (D)
- The structure of chlorophyll molecule 70. shows a porphyrin head and phytol tail. Porphyrin is a cyclic-tetra pyrrole ring to which phytol tail is attached. To which ring of porphyrin is the phytol tail attached?
 - (A) I
 - II ~ (B)
 - III (C)
 - (D) IV

- The entry of pollen tube through the 71. micropyle is:
 - (A) Allogamy
 - Geitonogamy (B)
 - Porogamy (C)
 - Chalazogamy ~ (D)
- Which disease of wheat is caused by Ustilago?
 - (A) Rust
 - Leaf spot / (B)
 - Smut (C)
 - Blast (D)
- determine homozygosity 73. heterozygosity, a plant must be crossed with:
 - Dominant parent (A)
 - Recessive parent (B)
 - Homozygous dominant (C)
 - (D) Heterozygous dominant
- Micropropagation is:
 - plants from (A) Production of Zoospores
 - Propagation of microorganisms (B)
 - Technique of obtaining new plants (C) by growing cells or tissues in culture medium
 - Technique of obtaining small plants (D)
- Exogenously borne non-motile asexual reproductive units in Ascomycetes are called:
 - Conidia (A)
 - Ascospore (B)
 - Basidiospore (C)
 - (D) Zoospores

- After fertilization the seed coat of a seed 76. develops from:
 - Embryo sac (A)
 - Integuments (B)
 - Nucellus (C)
 - Chalaza C (D)
- Which fungus was responsible for the 77. great Irish famine?
 - Leaf spot of rice (A)
 - Rust of wheat (B)
 - Powdery mildew of peas (C)
 - Late blight of potato 🗸 (D)
- Which effect does treatment of 78. gibberellins have on cabbage leaves?
 - Abscission
 - Bolting -(B)
 - Delayed senescence (C)
 - Early senescence (D)
- Which is unrelated pair? 79.
 - Gibberellins Gibberalla fujikuroi (A)
 - Ethylene Methionine (B)
 - Auxin Indole-3-acetic acid (C)
 - Cytokinin Violaxanthin
- 80. The phenomenon of single gene contributing to multiple phenotypic traits is called:
 - (A) Pleiotropy
 - (B) Multiple allelism
 - (C) Co-dominance
 - Polygenic inheritance

- **81.** To which class does Protopterus belong?
 - (A) Aves
 - (B) Reptilia /
 - (C) Amphibia
 - (D) Pisces
- **82.** Which stage of mitosis is regarded as the phase of reconstruction and reorganization of nucleus?
 - (A) Prophase
 - (B) Metaphase
 - (C) Anaphase V
 - (D) Telophase
- 83. In which type of linkage, chromosomes do not undergo any breakage during gametogenesis and no independent assortment occurs between the pairs of genes?
 - (A) Sex-linkage
 - (B) Inter-chromosomal linkage V
 - (C) Incomplete linkage
 - (D) Complete linkage 🗸

- 84. In humans the chromosomal composition of zygote destined to form a female is:
 - (A) 22 + X
 - (B) 22 + Y
 - (C) 44 + XX
 - (D) 44 + XY
 - 85. Where are the t-RNAs transcribed in the eukaryotes?
 - (A) Ribosome /
 - (B) Nucleolus
 - (C) ER
 - (D) Golgi body
 - 86. The walls and roof of a greenhouse out-door laboratory is made up of:
 - (A) Asbestus &
 - (B) GI sheet
 - (C) Glass
 - (D) Green grass

- 87. Which type of nutrition is seen in the animals who feed on their own faecal matter?
 - (A) Coprozoic
 - (B) Mesotrophic
 - (C) Osmotrophic
 - (D) Saprozoic
- 88. Which one is not a factor for formation of new species according to modern synthetic theory of evolution?
 - (A) Isolation
 - (B) Sexual selection
 - (C) Variation
 - (D) Natural selection
- 89. The phenomenon of industrial melanism was first observed in :
 - (A) USA
 - (B) United Kingdom
 - (C) Kenya
 - (D) Italy

- 90. Which one was responsible for Bhopal gas tragedy?
 - (A) Methane
 - (B) Methyl carbide
 - (C) Methyl isocyanate
 - (D) Melathion
- 91. The SA node is located in the:
 - (A) Right atrium
 - (B) Right ventricle
 - (C) Left atrium
 - (D) Left ventricle
- 92. In man, aerobic respiration is completed in :
 - (A) Nucleus
 - (B) Mitochondria
 - (C) Nucleolus
 - (D) Dictyosome
- 93. Which of the following statement, is true?
 - (A) All veins carry deoxygenated blood
 - (B) All arteries carry oxygenated blood ~
 - (C) All veins except one carry oxygenated blood
 - (D) All arteries except one carry oxygenated blood

- **94.** Which physiological process does occur inside the cytoplasm of a cell?
 - (A) Krebs Cycle
 - (B) Electron Transport
 - (C) Citric acid Cycle
 - (D) Glycolysis
- **95.** The pressure build up at the glomerular capillary network is called:
 - (A) Intrarenal pressure
 - (B) Hydrostatic pressure
 - (C) Osmotic pressure
 - (D) Interstitial pressure /
- **96.** Which one is an inhibitory neurotransmitter?
 - (A) Acetylcholine
 - (B) Histamine dopamine
 - (C) Gamma-aminobutyric acid
 - (D) Glutamate
- **97.** Ammonia is converted to urea through which cycle ?
 - (A) Citric acid Cycle ✓
 - (B) Cardiac Cycle
 - (C) Ornithine Cycle
 - (D) Sodium Co-transport

- 98. Which hormone is called 'love hormone'?
 - (A) Oxytocin
 - (B) Vasopressin ~
 - (C) Testosterone
 - (D) Melatonin
- **99.** Which one of the following enzymes is present in the acrosome of sperm head?
 - (A) Pepsinogen <
 - (B) Trypsin
 - (C) Hyaluronidase
 - (D) Isomerase
- 100. An abnormal condition in man where the testes fail to descend into scrotum and is retained in the condition called:
 - (A) Gubernaculum
 - (B) Cretinism
 - (C) Lorain dwarfism
 - (D) Cryptoorchism