

TS ECET - 2023

Syllabus for Metallurgical Engineering

MATHEMATICS (50 Marks)

Unit-I: Matrices

Matrices: Definition of Matrix, Types of matrices-Algebra of matrices-Transpose of a matrix-Symmetric, skew symmetric matrices-Minor, cofactor of an element-Determinant of a square matrix-Properties-Laplace's expansion-singular and non-singular matrices-Adjoint and multiplicative inverse of a square matrix-System of linear equations in 3 variables-Solutions by Cramer's rule, Matrix inversion method-Gauss-Jordan method.-Partial Fractions: Resolving a given rational function into partial fractions. Logarithms: Definition of logarithm and its properties, meaning of 'e', exponential function and logarithmic function.

Unit-II: Trigonometry

Properties of Trigonometric functions– Ratios of Compound angles, multiple angles, sub multiple angles – Transformations of Products into sum or difference and vice versa. Properties of triangles: sine rule, cosine rule, tangent rule and projection rule. Solution of a triangle when (i) three sides (SSS), (ii) two sides and an included angle (SAS), (iii) one side and two angles are given(SAA). Inverse Trigonometric functions, Hyperbolic functions.

Complex Numbers: Definition of a complex number, Modulus, amplitude and conjugate of complex number, arithmetic operations on complex numbers - Modulus-Amplitude form (Polar form) - Euler form (exponential form).

Unit-III: Analytical Geometry

Straight Lines–different forms of Straight Lines, distance of a point from a line, angle between two lines, intersection of two non-parallel lines and distance between two parallel lines. Circles-Equation of circle given center and radius, given ends of diameter-General equation- finding center and radius, center and a point on the circumference, 3 non-collinear points, center and tangent, equation of tangent and normal at a point on the circle. Conic Section – Properties of parabola, ellipse and hyperbola – Standard forms with vertex at origin.

Unit-IV: Differentiation and its Applications

Functions and limits – Standard limits – Differentiation of sum, product, quotient of functions, function of function, trigonometric, inverse trigonometric, exponential, logarithmic, Hyperbolic functions, implicit, explicit and parametric functions–Derivative of a function with respect to another function-Second order derivatives – Geometrical applications of the derivative (angle between curves, tangent and normal)–Increasing and decreasing functions–Maxima and Minima (single variable functions) using second order derivative only physical application – Rate Measure - Partial Differentiation–Partial derivatives up to second order–Euler's theorem.

Unit-V: Integration and its Applications

Indefinite Integral – Standard forms – Integration by decomposition of the integrand, integration of trigonometric, algebraic, exponential, logarithmic and Hyperbolic functions– Integration by substitution –Integration of reducible and irreducible quadratic factors – Integration by parts–Definite Integrals and properties, Definite Integral as the limit of a sum – Application of Integration to find areas under plane curves and volumes of Solids of revolution– Mean and RMS values, Trapezoidal rule and Simpson's 1/3 Rule for approximation integrals.

Unit-VI: Differential Equations

Definition of a differential equation-order and degree of a differential equation- formation of

differential equations-solution of differential equation of the type first order, first degree, variable-separable, homogeneous equations, exact, linear differential equation of the form $dy/dx+Py=Q$, Bernoulli's equation, 2nd order linear differential equation with constant coefficients both homogeneous and non-homogeneous and finding the Particular Integrals for the functions e^{ax} , $\sin ax$, $\cos ax$, ax^2+bx+c (a,b,c are real numbers).

Unit-VII: Laplace Transforms

Laplace Transforms (LT) of elementary functions-Linearity property, first shifting property, change of scale property, multiplication and division by t - LT of derivatives and integrals, Unit step function, LT of unit step function, second shifting property, evaluation of improper integrals, Inverse Laplace transform (ILT)-shifting theorems, change of scale property, multiplication and division by s , ILT by using partial fractions and convolution theorem. Applications of LT to solve linear ordinary differential equations up to second order with initial conditions.

Unit-VIII: Fourier Series

Fourier series, Euler's formulae over the interval $(C, C+2\pi)$ for determining the Fourier coefficients. Fourier series of simple functions in $(0, 2\pi)$ and $(-\pi, \pi)$. Fourier series for even and odd functions in the interval $(-\pi, \pi)$ - Half range Fourier series - sine and cosine series over the interval $(0, \pi)$.

PHYSICS (25 Marks)

Unit-I: UNITS, DIMENSIONS AND FRICTION

Physical quantity - Fundamental and derived quantities - Unit - definitions - S.I units - Advantages of S.I. units - Dimensions and dimensional formula - definitions-units and dimensional formula for physical quantities - Principle of homogeneity - Applications of dimensional analysis-Friction - causes - types of friction - Normal reaction - Laws of static friction - coefficients of friction - expression-rough horizontal surface - expressions for Acceleration, Displacement, Time taken to come to rest and Work done - Advantages and disadvantages of friction - Methods to reduce friction - Problems on friction only.

Unit-II: ELEMENTS OF VECTORS

Scalar and vector quantities - definitions and examples -Graphical representation of a vector - Classification of vectors (Proper vector, Unit vector, Equal vector, Negative vector, Collinear vector and Position vector) Resolution of a vector - Triangle law of vector addition - Parallelogram law of vectors - statement- expression for magnitude and direction of resultant vector -derivation-illustrations (working of sling and flying bird) - Representation of a vector in unit vectors \mathbf{i} , \mathbf{j} and \mathbf{k} - Scalar product of vectors-definition- application to work done by force - properties of scalar product - Vector product of vectors -definition - Right hand thumb rule and right hand screw rule - application to moment of force - properties of vector product - area of parallelogram and triangle in terms of vector product - related problems

Unit-III: MECHANICS

Projectile motion - definition - examples - Horizontal projection - Time of flight and Horizontal range - derivations - Oblique projection - Expression for path of a projectile in oblique projection - derivation- Maximum height, Time of ascent, Time of descent, Time of flight, Horizontal range and maximum horizontal range - derivations - Circular motion, angular velocity, time period and frequency of revolutions-Definitions- Relation between linear velocity and angular velocity - derivation-centripetal force - centrifugal force - definitions and expressions only- application

(banking of curved path) - angle of banking- expression only - related problems

Unit-IV: PROPERTIES OF MATTER

Elasticity and plasticity- definitions – examples - Stress and Strain – definitions and expressions - elastic limit - Hooke's law – statement - modulus of elasticity - Young's modulus – Derivation – Cohesive and adhesive forces - Surface tension - Illustrations - Capillarity –angle of contact – definition- examples for capillarity- Formula for Surface tension based on capillarity (no derivation) – Viscosity - Illustrations of viscosity - Newton's formula for viscous force – derivation - Coefficient of viscosity - Poiseuille's equation - Effect of temperature on viscosity of liquids and gases– streamlines - laminar flow - turbulent flow - Reynold's number - equation of continuity – statement - related problems.

Unit-V: HEAT AND THERMODYNAMICS

Heat – expansion of gases - Boyle's law –concept of absolute zero - Absolute scale of temperature – Charles' laws - Ideal gas equation – derivation - value of universal gas constant 'R' –Isothermal and Adiabatic processes - Differences between isothermal and adiabatic processes - Internal energy and external work done – Expression for work done – derivation – first law of thermodynamics – application of first law to isothermal and adiabatic processes - second law of thermodynamics – specific heat of a gas – molar specific heat of a gas – definitions – derive relation between C_p and C_v - related problems.

Unit-VI: CONSERVATION LAWS AND ENERGY SOURCES

Work and Energy - Potential Energy and kinetic energy–examples – expressions for PE and KE - derivations - Work-Energy theorem – derivation – Law of conservation of energy – examples - Law of conservation of energy in the case of freely falling body – proof – Illustration of conservation of energy in the case of simple pendulum– Non renewable and renewable energy sources - related problems

Unit-VII: WAVES AND SOUND

Wave motion – definition and characteristics – audible range – infrasonic and ultrasonic – longitudinal and transverse waves – examples – Relation between wavelength, frequency and velocity of a wave – derivation –stationary waves- beats - applications of beats - Doppler effect – list the applications – ultrasound and radar in medicine and engineering as special emphasis- echo – definition - applications - relation between time of echo and distance of obstacle –derivation- Reverberation and time of reverberation - Sabine's formula - Free and forced vibrations - Resonance - Conditions of good auditorium - noise pollution – definition – causes, effects and methods to minimize noise pollution - problems

Unit-VIII: SIMPLE HARMONIC MOTION

Periodic motion - Simple Harmonic Motion (SHM)– definition - examples - Conditions for SHM – Projection of circular motion on any diameter of a circle is SHM - Expressions for Displacement, Velocity and Acceleration of a particle executing SHM – derivations - Time period, frequency, amplitude and phase of a particle in SHM - Ideal simple pendulum – time period of simple pendulum –derivation - laws of simple pendulum-Seconds pendulum- problems

Unit-IX: MODERN PHYSICS

Photo electric effect - Einstein's photo electric equation – Work function and threshold frequency - laws of photo electric effect - applications of photo electric effect – photo cell - concept of Refraction of light - critical angle and total internal reflection - principle of Optical fiber -

Applications of optical fiber – LASER – definition and characteristics – principle of LASER - spontaneous emission and stimulated emission - population inversion - examples of LASER – Uses.

Unit-X: MAGNETISM

Magnetic field - magnetic lines of force -properties - Uniform and Non-uniform magnetic field – Magnetic length, pole strength – magnetic induction field strength- definition - Coulomb's inverse square law of magnetism - expression for moment of couple on a bar magnet placed in a uniform magnetic field – derivation - expression for magnetic induction field strength at a point on the axial line of a bar magnet –derivation - Dia, Para and Ferro magnetic materials – examples - related problems.

Unit-XI: ELECTRICITY AND MEASURING INSTRUMENTS

Ohm's law – Ohmic and non ohmic conductors – examples - Temperature dependence of resistance – coefficients of resistance with examples - Specific resistance – units – conductance- series and parallel combination of resistors - moving coil galvanometer - conversion of galvanometer into ammeter and voltmeter with diagram (qualitatively) – Kirchoff's current and voltage laws in electricity – Expression for balancing condition of Wheatstone's bridge – derivation – Meter bridge –working with neat diagram –Superconductivity-definition-superconductors - definition and examples – applications - related problems.

Unit-XII: ELECTRONICS

Solids – definition – energy bands in solids- valence band, conduction band and forbidden band – Energy band diagram of conductors, insulators and semiconductors – concept of Fermi level - Intrinsic semiconductors - examples - Concept of holes in semiconductors - Doping - Extrinsic semiconductor - P-type and N-type semiconductors - PN Junction diode – Forward Bias and Reverse Bias - Applications of PN diode - Diode as rectifier – principle – principle of Light Emitting Diode and solar cell.

CHEMISTRY (25 Marks)

Unit-I: Fundamentals of Chemistry

Atomic Structure: Introduction - Atomic number - Mass number- Bohr's Atomic theory - Aufbau principle - Hund's rule - Pauli's exclusion Principle- Orbitals, shapes of s, p and d orbitals - Electronic configuration of elements

Chemical Bonding: Introduction - Electronic theory of valency - Types of chemical bonds - Ionic, covalent and co-ordinate covalent bond with examples - Properties of Ionic and Covalent compounds

Oxidation-Reduction: Electronic Concepts of Oxidation-Reduction, Oxidation Number-calculations.

Unit -II: Solutions and Colloids

Introduction-Classification of solutions based on physical state- Molecular weights, Equivalent weights - Expression of concentration - Mole concept, Molarity, Normality, Numerical problems on Mole, Molarity and Normality - Colloids- Types of colloids- Lyophilic and Lyophobic colloids - Industrial applications of colloids.

Unit -III: Acids and Bases

Introduction - theories of acids and bases and limitations - Arrhenius theory-Bronsted-Lowry theory

- Lewis acid base theory - Ionic product of water - pH and related numerical problems - Buffer solutions- buffer action - applications of buffer solution.

Unit -IV: Environmental Studies-I

Introduction - environment -scope and importance of environmental studies- important terms - renewable and non-renewable energy sources - Concept of ecosystem, producers, consumers and decomposers - Biodiversity, definition and threats to Biodiversity- Forest resources- Over exploitation-Deforestation.

Unit -V: Water Technology

Introduction -soft and hard water - causes of hardness – types of hardness -disadvantages of hard water - degree of hardness (ppm) - softening methods - permutit process - ion exchange process - drinking water - municipal treatment of water for drinking purpose - Osmosis, Reverse Osmosis - advantages of Reverse Osmosis – Desalination by Electro dialysis - Defluoridation – Nalgonda technique.

Unit -VI: Electrochemistry

Conductors, insulators, electrolytes –Types of electrolytes - Arrhenius theory of electrolytic dissociation - electrolysis -electrolysis of fused NaCl and aqueous NaCl - applications of electrolysis - Faraday's laws of electrolysis- numerical problems.

Unit -VI I: Metallurgy

Characteristics of Metals - distinguish between Metals and Non Metals- Ore, Gangue, Flux and Slag - Concentration of Ore -Froth floatation - Methods of Extraction of crude Metal - Roasting, Calcination and Smelting - Alloys-purpose of making alloys - Composition of Brass, German silver, Nichrome, Stainless steel and Duralumin

Unit –VIII: Corrosion:

Introduction - factors influencing the rate of corrosion - electrochemical theory of corrosion - composition, stress and concentration cells- rusting of iron and its mechanism - prevention of corrosion - coating methods, Paints-constituents and characteristics of paints-cathodic protection

Unit –IX: Polymers

Introduction - polymerization - types of polymerization - addition, condensation with examples - plastics - types of plastics - advantages of plastics over traditional materials - Disadvantages of using plastics - preparation and uses of the following plastics: 1. Polythene 2. PVC 3.Teflon 4.Polystyrene 5.Urea formaldehyde 6. Bakelite - Rubber - Elastomers –Preparation of Butyl rubber, Buna-s rubber, Neoprene rubber and their uses-Fibres-Preparation and uses of fibres-Nylon 6,6- Polyester

Unit –X: Fuels

Definition and classification of fuels- characteristics of good fuel - Calorific value-HCV and LCV- Calculation of oxygen required for combustion of methane and ethane - composition and uses of gaseous fuels - a) Water gas b) Producer gas, c) Natural gas, d) Coal gas, e) Bio gas and f) Acetylene.

Unit –XI: Electro Motive Force

Galvanic cell – standard electrode potential -electro chemical series -emf of cell – Batteries-Types of batteries-Fuel cells.

Unit –XII: ENVIRONMENTAL STUDIES-II

Introduction- classification of air pollutants based on origin and states of matter - Air pollution; causes and effects - control methods - Water pollution; causes and effects - control measures.

METALLURGICAL ENGINEERING (100 Marks)**Unit-I: ELEMENTARY PRINCIPLES OF METALLURGY**

Introduction to metallurgy – ores & ore dressing, Methods of ore sampling – Comminution – sizing- concentration. Principles and processes of Pyro, hydro and electrometallurgy – Minerals of commercially important metals.

Unit-II: FUELS & REFRACTORIES

Classification of solid, liquid and gaseous fuels – Testing and properties of important fuels- Manufacture and characteristics of Metallurgical Coke – Gasification of solid fuels - Combustion of fuels – Properties, manufacture and selection of Refractories.

Unit-III: METALLURGICAL THERMODYNAMICS

Introduction and applications of thermodynamics – First Law of thermodynamics - Thermo chemistry - Second Law of thermodynamics - Ellingham diagrams - Fugacity, activity and equilibrium constant - Phase equilibria – Solutions.

Unit-IV: PHYSICAL METALLURGY

Structure of Metals and Alloys – Solidification – Binary thermal equilibrium diagrams - Iron-carbon diagram - important non- ferrous binary alloy systems – Microscopic and macroscopic examination of metals and alloys.

Unit-V: HEAT TREATMENT TECHNOLOGY

Heat treatment of plain carbon steels - Annealing, Normalizing, Hardening and tempering of steels – TTT diagrams – Hardenability - Grain size, Quenching media. Alloy steels & Effect of alloying elements on plain carbon steels – Stainless steels, tool steels – Case hardening techniques. Special heat treatment techniques such as Austempering, Martempering, sub-zero treatment - Heat treatment of Non-ferrous metals and alloys – Age hardening - Heat Treatment Furnaces and Furnace atmospheres – Heat treatment defects.

Unit-VI: FERROUS EXTRACTIVE METALLURGY

Iron ores and preparation of iron ores – Burden distribution – Blast furnace profile - Blast furnace plant and equipment – blast furnace reactions - irregularities and recent trends - sponge iron & methods of production – Ferroalloys – types and applications. Steel making by Bessemer, LD, Kaldo, Oxygen Lime (LDAC) process, Open hearth and Electric furnaces – Secondary steel making process – Vacuum treatment of liquid steel - Ingot defects - Continuous casting.

Unit-VII: NON-FERROUS EXTRACTIVE METALLURGY

Extraction of copper – Pyro and hydrometallurgical methods & refining – Aluminum - Extraction, Anode effect, Refining - Zinc and Lead - Pyro and hydrometallurgical extraction and refining. Extraction of Magnesium by Dows and pidgeon processes - Extraction of Titanium by Kroll's process - Refining of Titanium by Van arnell's process-Extraction of Thorium and Zirconium.

Unit-VIII: MATERIAL TESTING

Tension test. Stress - strain relationships, necking phenomenon. Hardness tests - principles and types. Impact testing - Notched bar impact tests. Transition temperature. Fatigue, Stress cycles, S-N diagram, Factors affecting Fatigue. Creep testing - creep curve, Stress - rupture test. Non-destructive testing- Principles, methods and applications of liquid penetrant, Radiography and Ultrasonic Magnetic particle.

Unit-IX: MECHANICAL METALLURGY

Plastic deformation of metals – lattice defects – Slip and Twinning - CRSS –Strengthening mechanisms. Strain hardening - Hot and Cold working - Recovery, recrystallisation and grain growth. Metal forming processes - Rolling, Forging, Extrusion & Sheet metal forming processes and defects.

Unit-X: FOUNDRY TECHNOLOGY

Patterns: Types, materials and pattern allowances, Moulding Sands - properties and

Testing, Moulding Processes and equipment: Sand casting, Die casting, Shell moulding, Centrifugal casting, Investment casting and CO₂ process-Cores: Types of Cores and properties, pouring and feeding of castings. Cast irons – types, Melting of cast irons - Grey, S.G and Malleable iron. Aluminum, Copper and Steel Foundry practices. Defects in Castings. Cleaning & Salvage of Castings.

Unit-XI: WELDING TECHNOLOGY

Basic concepts of Welding - Principles and processes of various welding techniques such as Oxy-acetylene, Shield Arc welding, Inert gas welding- TIG and MIG - Special welding processes - Plasma, resistance, electro slag, electron beam, thermit and Laser. Weld ability, factors affecting weld ability – Heat affected zone, Microstructure – Post weld treatments – Welding defects – Inspection and testing.
